

A Short Introduction to Chemical Biology and Medicinal Chemistry

Part II Ben Davis – 3 Lectures - Enzymes and Their Uses

Learning Outcomes

By the end of this you should be able to

- Compare and Contrast Advantages/Disadvantages of Enzymes as Reagents/Catalysts
- Recall the general mechanism of serine hydrolases
- Compare and contrast this mechanism with other peptide bond making/breaking catalysts
- Recall the mechanism of serine proteases, metalloproteases, aspartylproteases, the ribosome
- Describe examples of uses of acyl transferases in synthesis
- Explain the differing strategies used to accomplish bond making vs bond breaking using acyltransferases
- Give examples of the types of selectivities that may be exhibited by acyl transferases
- Explain the use of enzymes in kinetic resolution (KR), DKR, and desymmetrization
- Recall the mechanisms of glycosidases and glycosyltransferases
- Compare and contrast the mechanism of glycosidases with acyltransferases
- Give examples of methods for protein engineering
- Describe the basic principles behind site-directed mutagenesis
- Compare and contrast site-directed mutagenesis with chemical modification
- Give examples of protein engineering that has altered enzyme reactivity and specificity

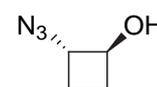
Enzyme movies and X-ray Structures

<http://www.chem.ox.ac.uk/researchguide/bgDavis.html>

Sample Exam Question Types

1. (a) Give an account of the use of serine hydrolase mechanism enzymes in synthesis [13 marks]
(b) Comment on the following transformations [5 x 4 marks]
Biocatalytic reaction e.g.s

2. (a) By giving examples of the way in which enzyme stereoselectivity can be exploited outline 3 complementary methods for obtaining an enantiopure sample of the following:



[15 marks]

3. (a) The enzyme *chuggase* has the following active site residues and operates on substrate **A** but not substrate **B**. By drawing analogies with classes of enzymes and biocatalysts with which you are familiar propose a mechanism for its catalytic cycle.

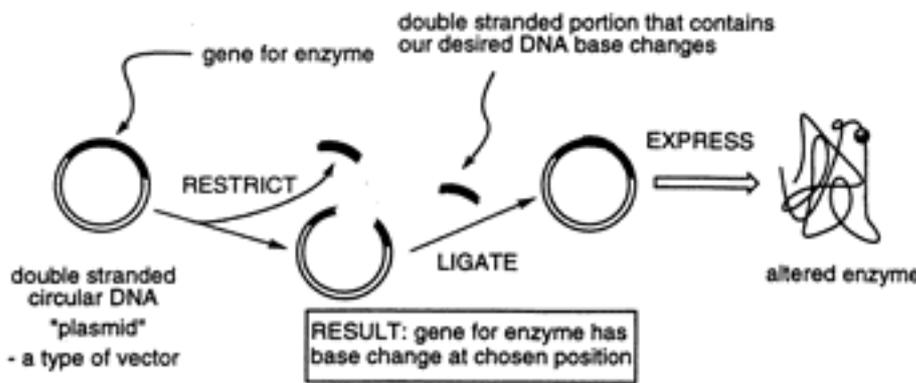


[15 marks]

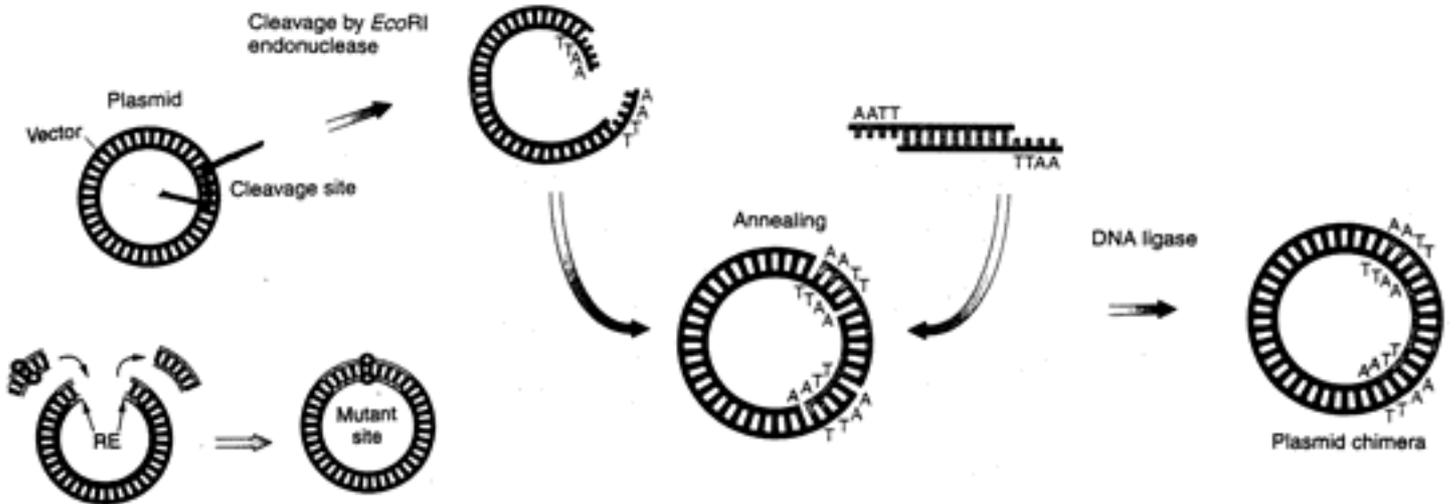
- (b) Given the mechanism that you have proposed, design a *chuggase* inhibitor.

[10 marks]

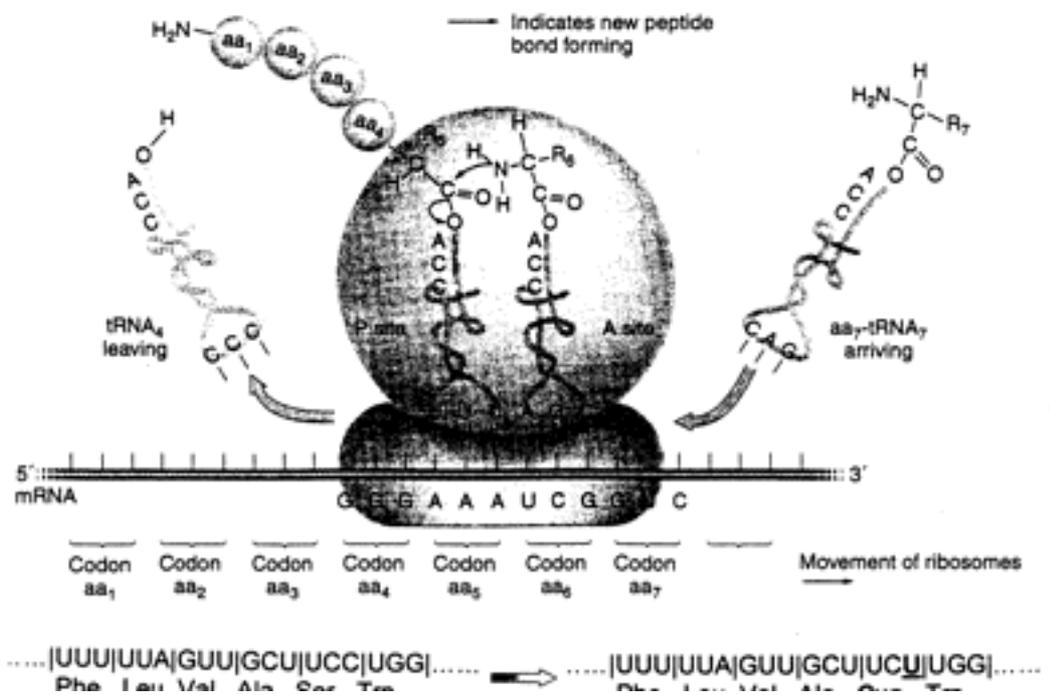
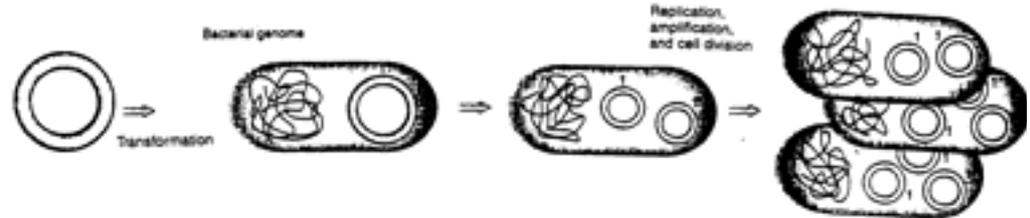
Site-Directed Mutagenesis



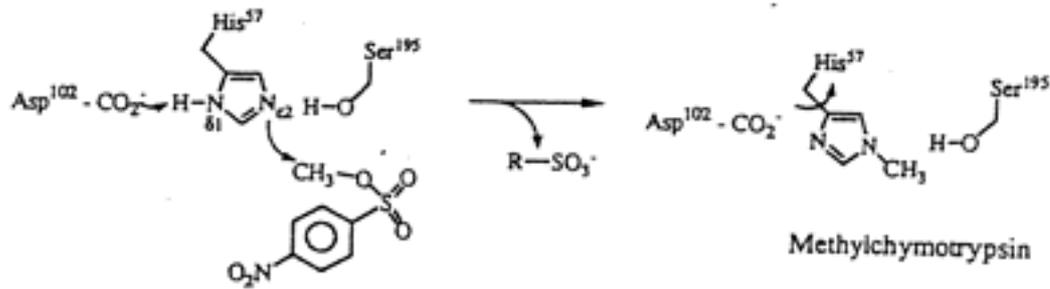
First letter	Second letter			
	U	C	A	G
U	UUU } Phe UUC } UUA } Leu UUG }	UCU } Ser UCC } UCA } UCG }	UAU } Tyr UAC } UAA } Stop UAG } Stop	UGU } Cys UGC } UGA } Stop UGG } Trp
C	CUU } Leu CUC } CUA } CUG }	CCU } Pro CCC } CCA } CCG }	CAU } His CAC } CAA } Gln CAG }	CGU } Arg CGC } CGA } CGG }
A	AUU } Ile AUC } AUA } AUG } Met	ACU } Thr ACC } ACA } ACG }	AAU } Asn AAC } AAA } Lys AAG }	AGU } Ser AGC } AGA } Arg AGG }
G	GUU } Val GUC } GUA } GUG }	GCU } Ala GCC } GCA } GCG }	GAU } Asp GAC } GAA } Glu GAG }	GGU } Gly GGC } GGA } GGG }



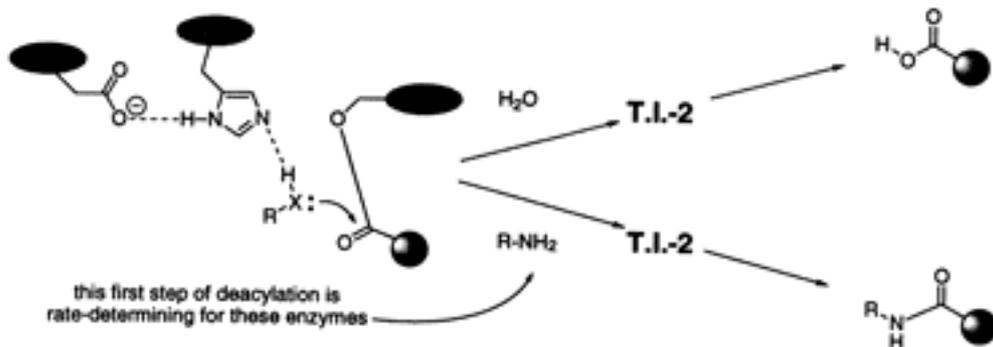
Enzyme	Source organism	Restriction site in double-stranded DNA
<i>EcoRI</i>	<i>Escherichia coli</i>	5' -G-A-A-T-T-C- 3' 3' -C-T-T-A-A-G- 5'
<i>EcoRII</i>	<i>E. coli</i>	5' -G-C-C-T-G-G-C- 3' 3' -C-G-G-A-C-C-G- 5'
<i>HindIII</i>	<i>Haemophilus influenzae</i>	5' -G-T-Py-Py-A-C- 3' 3' -C-A-Py-Py-T-G- 5'
<i>HindIII</i>	<i>H. influenzae</i>	5' -A-A-G-C-T-T- 3' 3' -T-T-C-G-A-A- 5'
<i>XbaI</i>	<i>H. aegyptius</i>	5' -G-G-C-C- 3' 3' -C-C-G-G- 5'
<i>XbaI</i>	<i>H. parainfluenzae</i>	5' -C-C-G-G- 3' 3' -G-G-C-C- 5'
<i>PstI</i>	<i>Providencia stuartii</i>	5' -C-T-G-C-A-G- 3' 3' -G-A-C-G-T-C- 5'
<i>SmaI</i>	<i>Serratia marcescens</i>	5' -C-C-C-G-G-G- 3' 3' -G-G-G-C-C-C- 5'
<i>BamI</i>	<i>Bacillus amyloliquefaciens</i>	5' -G-A-T-C-C- 3' 3' -C-C-T-A-G-G- 5'
<i>BglII</i>	<i>E. globosus</i>	5' -A-G-A-T-C-T- 3'



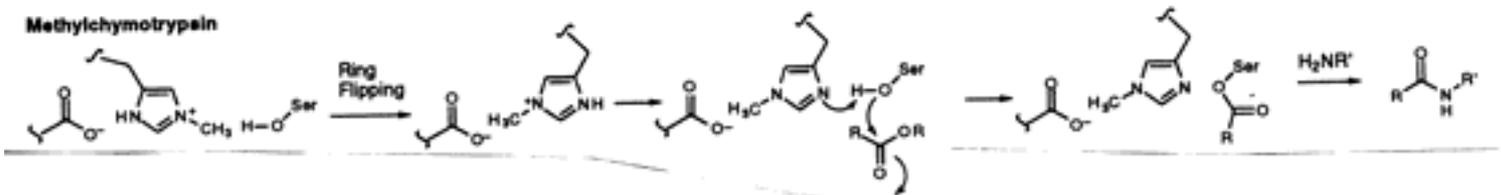
Methylchymotrypsin



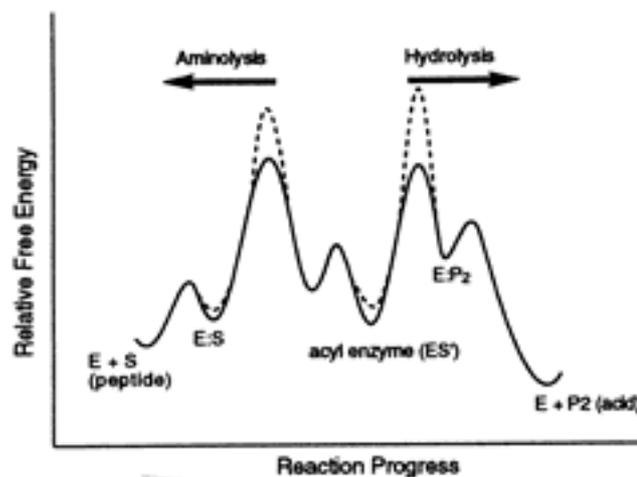
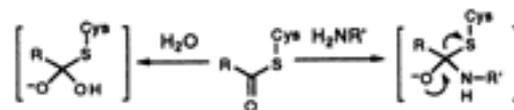
Altered Synthetic Properties of MethylCT and Thio/Selenosubtilisin



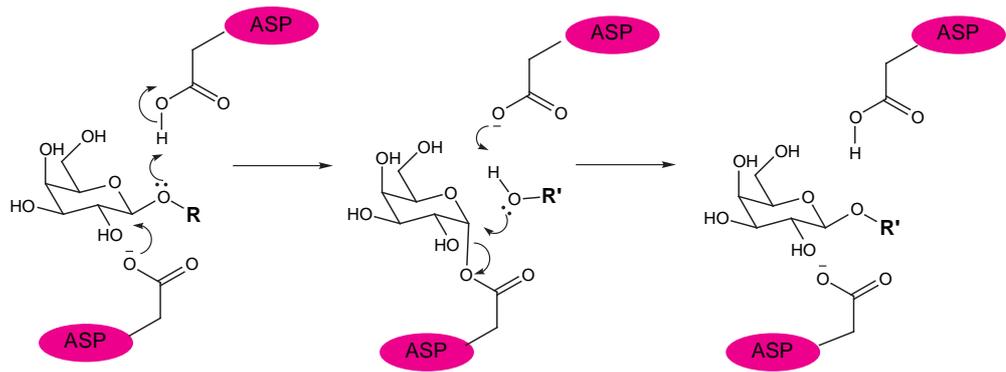
Methylchymotrypsin



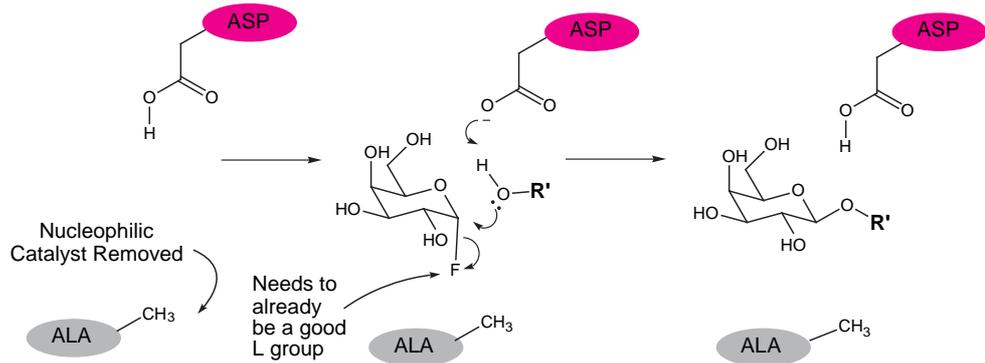
Thiosubtilisin



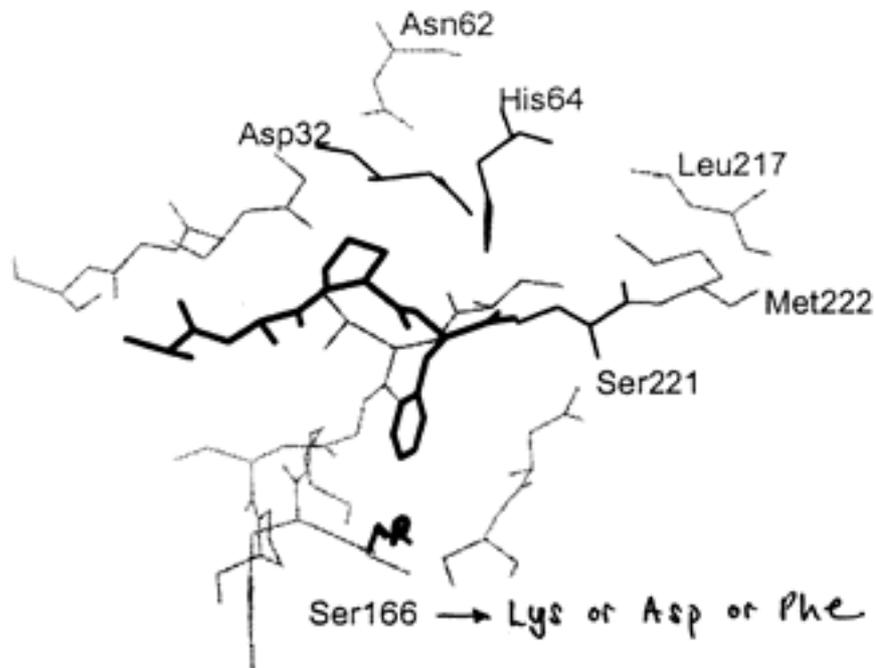
Glycosynthases
Normal β -galactosidase



A " β -galactosynthase"



Engineering Specificity



Amino Acid	Side Chain -CH ₂ R	$k_{cat}/K_M^* / s^{-1}mM^{-1}$			
		Phe	Lys	Asp	Ala
Ser	OH	200	120	110	30
Lys	(CH ₂) ₃ NH ₃ ⁺	10	0.2	240	40
Asp	COOH	47	280	1.2	47
Phe	C ₆ H ₅	5	19	40	180

* k_{cat}/K_M is known as the specificity constant as it provides a great idea of the rate of reaction of two competing substrates